α -N-Methyl/phenyl furan derivatives as dipolar ophiles for synthesis of spiro isoxazolidine derivatives with α -chloro and simple nitrones

Bhaskar Chakraborty*, Prawin Kumar Sharma, Neelam Rai, Sauray Kafley and Manjit Chhetri

Organic Chemistry Laboratory, Sikkim Government College, Gangtok, Sikkim-737102, India

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1,3-Dipolar cycloaddition reactions of α -chloro and simple nitrones have been studied with novel α -N-methyl/phenyl furan derivatives as new dipolarophiles. The reactions are found to be highly regioselective to afford single 5-spiro isoxazolidines with high yield in a short reaction time.

Keywords: α-N-methyl/phenyl furan derivatives, spiro cycloadducts, regioselectivity

In addition to the reported dipolar ophiles for 1,3-dipolar cycloaddition reactions of nitrones,1 we now describe some further novel and efficient dipolarophiles (α-N-methyl/phenyl furan derivatives) for the cycloaddition reactions to afford solely 5-spiro isoxazolidines with high yield in a very short reaction time (4-6 h) with different nitrones at RT (Scheme 1). A detailed literature survey reveals that these type of cycloaddition reactions are generally diastereomeric in nature with the predominance of one of the isomers.2 The synthetic potential of α-chloro nitrones (1) has been proved not only in cycloaddition reactions but also in the synthesis of novel dipolarophiles (2) when treated with alkyl halides.³ α -N-methyl/phenyl furan derivatives (2) were isolated as side-products and in single E isomeric forms (almost 20%) in the reported oxidation reaction of alkyl halides to aldehydes and ketones using α-chloro nitrones^{4–8} (Scheme 2). The novel 5-spiro isoxazolidines (**3–8**) were predominantly obtained regioselectively in high yields (78–88%) as single isomers in the reactions of α -chloro, α -amino and simple nitrones. This could be due to the fact that the nitrone (LUMO)-dipolarophile (HOMO) interactions are strong enough to dominate the reaction9-11 and lead to the formation solely of 5-spiro cycloadducts (3–8) via an exo approach of the nitrone 1 (all the reported nitrones have a Z configuration) to the furan derivatives 2 (transition state 1). At the outset of this work it was unclear whether the side-products obtained during aldehyde and ketone synthesis could be employed as efficient dipolarophiles. For the present study, we have used five different nitrones: N-methyl-α-chloro nitrone,⁶ N-phenylα-chloro nitrone,⁸ N-methyl-α-amino nitrone,¹² N-phenyl-αamino nitrone^{13–15} and N-methyl/phenyl nitrones¹⁶ respectively in order to generalise regioselectivity in cycloaddition reactions using novel dipolarophiles (2). The stereochemistry of the isoxazolidine derivatives (3-8) were rationalised by considering the multiplicity of the proton signals at 3-H, 4-H, CHCl (in the case of α -chloro nitrones only) asymmetric

centres along with their coupling constant values. 17,18 In the ¹H NMR spectrum of cycloadducts 3–4, 3-H resonates around δ_{H} 2.50–3.50 ppm while 4-H around $\delta_{\rm H}$ 3.00–5.85 ppm and the coupling constant is $J_{3,4} \sim 9.16$ Hz implying a cis relationship between H-3 and H-4. The CHCl proton also resonates upfield around $\delta_{\rm H} 2.20-2.60$ ppm. The 3-H and CHCl protons are also syn as evidenced from their coupling constant values $(J_{3,CHC})^{-1}$ 9.40 Hz).^{17,18} Almost similar coupling constant values are obtained for H-3 and H-4 protons in the case of other reported cycloadducts (5-8). Cycloaddition of Z nitrone via exotransition state geometry results in syn spiro isoxazolidine derivatives. The ¹H NMR spectra of 3–8 also show significant long range coupling between H-4 with H-3' and vice-versa in most of the spiro cycloadducts. In the mass spectra, in addition to molecular ion peaks, prominent base peak values are obtained in all the cycloadducts and significant M+2 peaks of characteristic relative heights are also obtained in 3–4 which is due to the isotopic abundance of Cl³⁷ atoms. Studies of HRMS spectra show almost exact masses for the majority of the compounds. The experimental procedure is very simple. α-N-methyl/phenyl furan derivatives (2) are added to nitrone 1 in diethyl ether at RT. Smooth reaction ends with the production of the novel spiro cycloadducts with extremely good yield in a very short reaction time. In general, the reactions are very clean and high yielding compared to other cycloaddition reactions of α -chloro and α -amino nitrones. 1,3-8,13-15 The products are characterised from their spectroscopic (IR, ¹H NMR, HRMS, ¹³C NMR) data. No catalyst or co-organic solvent is required. All the novel isoxazolidine derivatives (3-8) were also screened for antibacterial activity and found to be very active.

A preferential conformation for the spiro cycloadducts (3–8) is shown in Fig. 1.

A new mechanistic pathway for the synthesis of novel dipolarophiles (α -N-methyl/phenyl furan derivatives) may be represented in the following mechanism³ (Scheme 2).

H

$$R^{1}$$
 R^{1}
 R^{1}

 $R = CH_3 ; C_6H_5$

 $R^1 = C_6H_5$; CHCl(CH₂)₃OH; NH₂

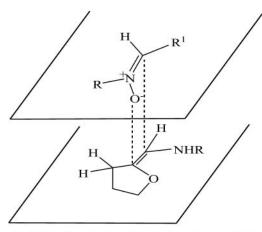
Reagents and conditions: i) Dry ether, RT, N2 atmosphere, 4 - 6 hr

Scheme 1

^{*} Correspondent. E-mail: bhaskargtk@yahoo.com

 $R^1 = Ph, NH_2, CHCl(CH_2)_3OH$

General conformation for the spirocycloadducts (3-8)



TS 1 for forming spiro cycloadducts (3-8)

Reagents and conditions: i) Dry ether, pyridine, r.t, N2 atmosphere

ii) Dry ether, Na₂CO₃, r.t , N₂ atmosphere

Scheme 2

Finally, we have reported the synthesis of novel spiro cycloadducts using novel dipolarophiles with different nitrones in 1,3-dipolar cycloaddition reaction. The formation of the desired compounds was obtained in good yields within a short reaction time. The notable advantages offered by this method are one-pot synthesis, simple operation, easy workup, mild and faster reaction conditions with high yield of products.

Experimental

¹H NMR spectra were recorded with a Bruker Avance DPX 300 spectrometer (300 MHz, FT NMR) using TMS as internal standard. 13C NMR spectra were recorded on the same instrument at 75 MHz. The coupling constants (J) are given in Hz. IR spectra were obtained with a Perkin-Elmer RX 1-881 as film or as KBr pellets for all the products. MS spectra were recorded with a Jeol SX-102 (FAB) instrument. The HRMS spectra were recorded on a Q-Tof micro instrument (YA-105).

TLC's were run on Fluka silica gel precoated TLC plates while column chromatography was performed with silica gel (E.Merck India) 60-200 mesh. All the alkyl halides, reagents and solvents were purified after receiving from commercial suppliers. N-methylhydroxylamine was purchased from Aldrich Chemical Company and was used as received. N-Phenylhydroxylamine was prepared following standard methods available in the literature and has been used already for the reported synthesis of aldehydes and in cycloaddition reactions involving α -amino, α -chloro nitrones in aqueous phase and in organic solvents. ^{3-8,12-15}

General procedure for synthesis of aldehyde and furan derivatives 2 To a stirred solution of nitrone 1 (R=Me; 500mg, 3.0198 mmol) in dry ether (25 ml) was added pyridine (1 equivalent) and stirred at RT with a magnetic stirrer under N₂ atmosphere for 1 h while the formation of transient nitrone **1a** (not isolated) was monitored by TLC ($R_f = 0.38$). Benzyl chloride (292.mg, 1 equiv) was added at this stage and the reaction mixture was stirred for another 3 h until the intermediate

compound 1b (not isolated) was developed (monitored by TLC; R,= 0.40). Solid Na₂CO₂ (2g.) was added at this stage and the reaction mixture was stirred for a further 1 h while the progress of the reaction was again monitored by TLC ($R_c = 0.43, 0.50$). During this process the N-O bond was cleaved in the basic medium. 19 Basic workup, removal of pyridine hydrochloride and silica gel column chromatographic purification using ethyl acetate-hexane provided the desired benzaldehyde as a colourless liquid (702 mg, 82%) and α-N-phenyl furan derivative (2) as a pale yellow gummy liquid (88mg, 17%). This procedure was followed for the synthesis of α -N-methyl furan derivative with other alkyl halides.

General procedure for cycloaddition (for regioselective spiro cycloadducts)

To a well stirred solution of nitrone 1 (R=Me; 1 mmole) in diethyl ether (20 mL) taken in a 50 mL conical flask, was added the α-N-methyl furan derivative [(E)-1-(dihydrofuran-2-(3H)-ylidene)-N-methyl methanamine)] (1 equiv) which was stirred at RT with a magnetic stirrer under N, atmosphere for 4 h. The progress of the reaction was monitored by TLC ($R_c = 0.53$). After completion of the reaction and work-up, the crude spiro-cycloadduct was concentrated in a rotary evaporator and finally purified by column chromatography using ethyl acetate - hexane to afford pure spiro-cycloadduct 3 (entry 1, Table 1, Scheme 1). This procedure was followed for the reaction of nitrone 1 (R= Me,Ph) with the α -N-methyl/phenyl furan derivatives 2 [(E)-1-(dihydrofuran-2-(3H)-ylidene)-N-methyl methanamine)/(E)-1-(dihydrofuran-2-(3H)-ylidene)-N-phenyl methanamine)] listed in Table 1.

 $(R=Me; \alpha-N-methyl furan derivative) [(E)-1-(dihydrofuran-2-(3H)$ ylidene)-N-methyl methanamine)] (2): IR (KBr): 3125-3054 (br), 2838 (m), 1652 (s), 1455 (m), 1210 (m) cm⁻¹; ¹H NMR (CDCl₂): δ 4.81 (br, 1H, N-H), 4.56 (s, 1H, C=CH), 3.30 (N-Me), 2.50–2.16 (m, 6H); ¹³C NMR (CDCl₃): δ 103.00, 101.76 (double bonded carbons), 26.22, 25.30, 23.65 (3 CH₂ carbons); FAB-MS: m/z 113 (M⁺), 98, 97; HRMS-EI: Calcd for C₆H₁₁ON (M), 113.1000. Found: M+, 112.9876.

 $(R=Ph; \alpha-N-phenyl furan derivative) [(E)-1-(dihydrofuran-2-(3H)$ ylidene)-N-phenyl methanamine)] (2): IR (KBr): 3150-3060 (br), 2860 (m), 1640 (s), 1430 (m), 1140 (m), 778 (s) cm⁻¹; ¹H NMR (CDCl₃): δ 7.83 (m, 5H, C_H₅), 6.29 (br, 1H, N-H), 2.17 (s, 1H, C=CH), 1.79–1.18 (m, 6H); ^{1.3}C NMR (CDCl₃): δ 137.20, 135.65, 134.00, 132.15 (aromatic carbons), 106.24, 104.18 (double bonded carbons), 28.46, 27.10, 24.84 (3 CH, carbons). FAB-MS (m/z): 175 (M⁺), 98, 97, 77. HRMS-EI: Calcd for C₁₁H₁₃ON (M), 175.0993. Found: M+, 175.0981.

(S)-4-Chloro-4-((3S,4S,5R)-2methyl-4-(methylamino)-1,6-dioxa-2azaspiro[4.4]nonan-3-yl)butan-1-ol (3): Pale yellow gummy liquid. Yield 88%, $R_c = 0.53$; IR (KBr): 3460–3326 (br), 2948 (m), 2420 (m), 1485 (s), 1325 (m), 810 (m), 774 (s) cm⁻¹; ¹H NMR (CDCl₃): δ 4.83 (br, 1H, CH₂OH, exchanged in D₂O), 4.60 (s, 1H, NHCH₂), 3.37

Table 1 Cycloaddition reaction of nitrone 1 with novel dipolar philes (2)

Entry	Nitrone (1)	Novel dipolarophiles ^a (2)	Spiro cycloadducts ^b (3–8)	Time /h	Nature of products	Yield /%°
1	$R = Me$ $R^1 = CHCI(CH_2)_3OH$	NHCH ₃	H H Me 3	4	Pale yellow gummy liquid	88
2	$R = Ph$ $R^1 = CHCI(CH_2)_3OH$	NHC6H3	O H H C ₆ H ₅ (CH ₂) ₃ OH H C ₆ H ₅ 4	5	Dark red viscous liquid	86
3	$R = Me$ $R^1 = NH_2$	$\bigcap_{\text{O}} \bigoplus_{\text{NHCH}_3}^{\text{H}}$	O H N H Me H 5	5	Grey viscous liquid	84
4	$R = Ph$ $R^1 = NH_2$	NHC ₆ H ₅	C ₆ H ₅ H C ₆ H ₅ 6	6	Dark grey viscou liquid	s 81
5	R = Me R¹ = Ph	NHCH ₃	CH ₃ C ₆ H ₅ H Me 7	5	Colourless gummy liquid	78
6	$R = R^1 = Ph$	$\bigcup_{O} \overset{H}{\underset{NHC_{6}H_{5}}{\bigvee}}$	H N H C ₆ H ₅	6	Colourless gummy liquid	78

 $^{^{}m e}$ Reaction conditions: nitrone (1 mmole), furan derivative (1 equiv),dry ether, N $_{
m e}$ atmosphere, RT.

^b All the spiro cycloadducts were characterised by IR, ¹H NMR, MS, ¹³C NMR spectral data.

clsolated yield after purification.

(s, 6H, 2 x N-CH₂), 3.12 (dd, 1H, J = 9.20, 8.32 Hz, C₂H), 2.70 (dt, 1H, J = 8.10, 7.88 Hz, C₄H), 2.35 (dt~m, 1H, CHCl), 1.88–1.42 (m, 6H); ¹³C NMR (CDCl₃): δ 93.00 (CHCl), 87.55 (C₅), 76.20 (C₃), 55.20 (C₄), 41.97 (N-CH₂), 40.24 (NH-CH₂), 33.37, 31.50, 28.68, 26.00, 25.12, 23.40 (6 CH₂ carbons); MS (m/z): 280 (M+2), 278 (M+), 263, 248, 156 (B.P), 141, 107; HRMS-EI: Calcd for C₁₃H₂₃O₃N₂Cl (M), 278.6710. Found: M+, 278.6698.

 $(S)\hbox{-}4-Chloro\hbox{-}4-((3S,4S,5R)\hbox{-}2-phenyl\hbox{-}4-(phenylamino)\hbox{-}1,6-dioxa$ 2-azaspiro[4.4]nonan-3-yl)butan-1-ol (4): Dark red viscous liquid. Yield 86%, $R_c = 0.48$; IR (KBr): 3485–3290 (br), 2962 (m), 2425 (m), 1620 (s), 1490 (s), 1260 (m), 1040 (m), 780 (s) cm⁻¹; ¹H NMR $(CDCl_3)$: δ 6.98 - 6.92 (m, 10H, 2 x C_6H_5), 5.84 (dd, 1H, J = 8.55, 8.20 Hz, C_3H), 5.00 (br, 1H, CH₂OH, exchanged in D_2O), 3.60 (dt, 1H, J =9.34, 7.88 Hz, C₄H), 3.40 (s, 1H, N–H proton of NHPh), 2.68 (dt~m, 1H, CHCl), 1.90 (dt, 1H, J = 6.82, 6.64 Hz, C_3 H), 1.50–1.12 (m, 4H); ¹³C NMR (CDCl₂): δ 138.00, 136.50, 134.30, 133.80, 131.75, 130.42, 129.46, 128.64 (aromatic carbons), 95.10 (CHCl), 86.40 (C₅), 73.75 (C₂), 53.30 (C₄), 30.20, 28.55, 27.34, 26.22, 25.73, 24.37 (6 CH₂) carbons); MS (*m/z*): 404 (M⁺+2), 402 (M⁺), 325, 310, 309, 218 (B.P), 107, 91, 77. HRMS-EI: Calcd for C₂₂H₂₇O₃N₂Cl (M), 402.7130. Found: M+, 402.7122.

(S)-3-Amino-(3S,4S,5S)-2-methyl-4-(methylamino)-1,6-dioxa-2azaspiroisoxazole (5): Grey viscous liquid. Yield 84%, R_s = 0.52; IR (KBr): 3430–3380 (br), 3033 (m), 2955 (m), 1773 (s), 1662 (s), 1480 (s), 1282 (m), 1178 (s), 806 (s) cm⁻¹; ¹H NMR (CDCl₃): δ 4.90 (br,s, 2H, NH₂, exchanged in D₂O), 4.60 (br, 1H, NHCH₃), 3.36 (s, 6H, 2 x N-CH₂), 3.00 (d, 1H, J = 7.54 Hz, C_3H), 2.70 (dt, 2H, J = 6.24, 6.28Hz, $C_3^{''}$ 2H), 2.38 (dt, 1H, J = 7.12, 6.70 Hz, C_4 H), 1.70–1.48 (m, 4H); 13 C NMR (CDCl₃): δ 88.50 (C_5/C_2), 77.12 (C_3), 56.26 (C_4), 40.94 (N-CH₃), 38.13 (NH-CH₃), 32.07, 31.22, 29.34 (3',4',5' CH, carbons); MS (m/z): 187 (M+), 172, 157, 156 (B.P), 141. HRMS-EI: Calcd for C₈H₁₇O₂N₃ (M), 187.1633. Found: M⁺, 187.1613.

(S)-3-Amino-(3S,4S,5S)-2-phenyl-4-(phenylamino)-1,6-dioxa-2azaspiroisoxazole (6): Dark grey viscous liquid. Yield 81%, $R_r = 0.48$; IR (KBr): 3436–3390 (br), 3030 (m), 2952 (m), 1780 (s), 1674 (s), 1480 (m), 1276 (m), 815 (s) cm⁻¹; ¹H NMR (CDCl₂): δ 7.02–6.90 (m, 10H, 2 x C_6H_5), 5.86 (d, 1H, J = 6.30 Hz, C_3H), 5.00 (br,s, 2H, NH₂), exchanged in D_2O), 3.50 (dt, 2H, J = 6.74, 6.06 Hz, C_3 2H), 3.38 (br,s,1H, NHC₆ \tilde{H}_5), 2.70 (dt, 1H, J = 7.20, 6.18 Hz,C₄H), 1.52–1.28 (m, 4H); ¹³C NMR(CDCl₂): δ 137.21, 135.44, 134.00, 133.10, 130.66, 129.40, 128.32, 127.84 (aromatic carbons), 86.94 (C₂/C₂), 74.24 (C₂), 55.70 (C₄), 27.87, 25.63, 24.00 (3',4',5' CH, carbons); MS (m/z): 311 (M+), 295, 218, 203 (B.P), 202, 92, 77; HRMS-EI: Calcd for C₁₈H₂₁O₂N₃ (M), 311.2054. Found: M⁺, 311.2037.

(S)-3-phenyl-(3S,4S,5S)-2-methyl-4-(methylamino)-1,6-dioxa-2azaspiroisoxazole (7): Colourless gummy liquid. Yield 78%, R_s = 0.52; IR (KBr): 3040 (m), 2965 (m), 1760 (s), 1685 (m), 1464 (s), 1290 (m), 1084 (s), 808 (m) cm⁻¹; ¹H NMR (CDCl₂): δ 6.81(s,5H,C H₅), 4.67 (s, 1H, NHCH₃), 3.36 (s, 6H, 2 x N-CH₃), 3.00 (d, 1H, $J = 5.74 \text{ Hz}, C_3 \text{H}), 2.74 \text{ (dt, 1H, } J = 6.64, 6.30 \text{ Hz}, C_4 \text{H}), 2.30 \text{ (dt, 2H, }$ J = 5.10, 4.92 Hz, C₃'2H), 1.80–1.55 (m, 4H); ¹³C NMR (CDCl₃): δ 129.05, 128.53, 128.27, 127.22 (aromatic carbons), 80.28 (C_5/C_2), 70.36 (C₂), 59.70 (C₄), 45.17 (N-CH₂), 41.64 (NH-CH₂), 32.07, 31.22, 29.34 (3',4',5'CH, carbons); MS (m/z):248 (M+), 218, 171, 156 (B.P), 141, 77. HRMS-EI: Calcd for C₁₄H₂₀O₂N₂ (M), 248.1862. Found: M+, 248.1853.

(S)-3-Phenyl-(3S,4S,5S)-2-phenyl-4-(phenylamino)-1,6-dioxa-2azaspiroisoxazole (8): Colourless gummy liquid. Yield 78%, R_e = 0.48; IR (KBr): 3024 (m), 2950 (m), 1772 (s), 1670 (s), 1468 (m), 1382 (m), 805 (m), 780 (s) cm⁻¹; ¹H NMR (CDCl₃): δ 7.50–6.62 (m, 15H, 3 x C₆H₅), 5.84 (s,1H, NHC₆H₅), 4.63 (d, 1H, J = 6.06 Hz, C₃H), 4.02 (dt, 1H, J = 6.18, 6.20 Hz, $C_{A}H$), 2.64 (dt, 2H, J = 5.28, 4.10 Hz,C₃'2H), 2.00–1.26 (m, 4H); ¹³C NMR (CDCl₃): δ 136.76, 136.53, 136.24, 135.15, 134.90, 134.62, 134.30, 133.78, 132.44, 132.18, 130.92, 130.37 (aromatic carbons), 83.22 (C₅/C₂), 71.52 (C₃), 52.89 (C₄), 23.61, 22.57, 21.14 (3',4',5' CH₂ carbons); MS (m/z): 372 (M⁺), 295, 280, 218, 203 (B.P), 92, 77; HRMS-EI: Calcd for C₂₄H₂₄O₂N₂ (M), 372.2286. Found: M+, 372.2270.

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